

(1*E*,4*E*)-1-(3-Nitrophenyl)-5-phenylpenta-1,4-dien-3-oneS. Samshuddin,^a Ray J. Butcher,^b Mehmet Akkurt,^{c*}
B. Narayana,^a B. K. Sarojini^d and H. S. Yathirajan^e^aDepartment of Studies in Chemistry, Mangalore University, Mangalagangotri, Mangalore 574 199, India, ^bDepartment of Chemistry, Howard University, 525 College Street NW, Washington, DC 20059, USA, ^cDepartment of Physics, Faculty of Sciences, Erciyes University, 38039 Kayseri, Turkey, ^dDepartment of Chemistry, P. A. College of Engineering, Nadupadav, Mangalore-574 153, India, and ^eDepartment of Studies in Chemistry, University of Mysore, Manasagangotri, Mysore 570 006, India
Correspondence e-mail: akkurt@erciyes.edu.tr

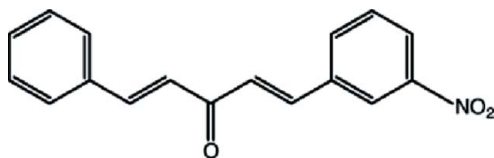
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Key indicators: single-crystal X-ray study; $T = 293\text{ K}$; mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$; R factor = 0.055; wR factor = 0.128; data-to-parameter ratio = 19.7.

In the title compound, $\text{C}_{17}\text{H}_{13}\text{NO}_3$, the dihedral angle between the benzene rings is $31.21(5)^\circ$. In the crystal, inversion dimers linked by pairs of $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds occur. A $\text{C}-\text{H}\cdots\pi$ interaction is also indicated.

Related literature

For the pharmacological importance of chalcones and bis chalcones, see: Sarojini *et al.* (2006); Dhar (1981); Dimmock *et al.* (1999); Satyanarayana *et al.* (2004). For our work on synthesis of different derivatives of chalcones, see: Baktir *et al.* (2011); Fun *et al.* (2010); Jasinski *et al.* (2010); Samshuddin *et al.* (2011*a,b,c*). For related structures, see: Butcher *et al.* (2006*a,b*; 2007*a,b,c*); Harrison *et al.* (2006); Hu *et al.* (2004); Fischer *et al.* (2007); Patil *et al.* (2007); Zhao *et al.* (2007).

**Experimental***Crystal data*

$\text{C}_{17}\text{H}_{13}\text{NO}_3$	$V = 1349.21(14)\text{ \AA}^3$
$M_r = 279.28$	$Z = 4$
Monoclinic, $P2_1/c$	Mo $K\alpha$ radiation
$a = 11.9806(6)\text{ \AA}$	$\mu = 0.10\text{ mm}^{-1}$
$b = 9.8955(4)\text{ \AA}$	$T = 293\text{ K}$
$c = 12.5562(7)\text{ \AA}$	$0.44 \times 0.34 \times 0.08\text{ mm}$
$\beta = 114.992(7)^\circ$	

Data collection

Oxford Diffraction Xcalibur Ruby Gemini diffractometer	7835 measured reflections
Absorption correction: multi-scan (<i>CrysAlis RED</i> ; Oxford Diffraction, 2007)	3746 independent reflections
$T_{\min} = 0.969$, $T_{\max} = 1.000$	2453 reflections with $I > 2\sigma(I)$
	$R_{\text{int}} = 0.033$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.055$	190 parameters
$wR(F^2) = 0.128$	H-atom parameters constrained
$S = 1.02$	$\Delta\rho_{\max} = 0.25\text{ e \AA}^{-3}$
3746 reflections	$\Delta\rho_{\min} = -0.26\text{ e \AA}^{-3}$

Table 1Hydrogen-bond geometry (\AA , $^\circ$).

Cg1 is the centroid of the C12–C17 phenyl ring.

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C2}-\text{H2A}\cdots\text{O3}^i$	0.93	2.59	3.412 (2)	147
$\text{C5}-\text{H5A}\cdots\text{Cg1}^{ii}$	0.93	2.62	3.3915 (19)	141

Symmetry codes: (i) $-x + 1, -y, -z + 1$; (ii) $-x + 1, y + \frac{1}{2}, -z + \frac{3}{2}$.

Data collection: *CrysAlis PRO* (Oxford Diffraction, 2007); cell refinement: *CrysAlis PRO*; data reduction: *CrysAlis RED* (Oxford Diffraction, 2007); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *ORTEP-3* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999), *PARST* (Nardelli, 1983) and *PLATON* (Spek, 2009).

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Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: TK5031).

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supporting information

Acta Cryst. (2012). E68, o74–o75 [doi:10.1107/S1600536811052548]

(1*E*,4*E*)-1-(3-Nitrophenyl)-5-phenylpenta-1,4-dien-3-one

S. Samshuddin, Ray J. Butcher, Mehmet Akkurt, B. Narayana, B. K. Sarojini and H. S. Yathirajan

S1. Comment

Chalcones are highly reactive substances of varied nature. They have been reported to possess many interesting pharmacological activities (Dhar, 1981) including anti-inflammatory, antimicrobial, antifungal, antioxidant, cytotoxic, antitumor and anticancer activities (Dimmock *et al.*, 1999; Satyanarayana *et al.*, 2004). Chalcones are also finding application as organic nonlinear optical materials (NLO) for their SHG conversion efficiency (Sarojini *et al.*, 2006). The basic skeleton of chalcones which possess α,β -unsaturated carbonyl group is useful for the synthesis of various biodynamic cyclic derivatives such as pyrazoline, benzodiazepine, 2,4,6-triaryl pyridine, isoxazoline and cyclohexenone derivatives (Samshuddin *et al.*, 2011*a,b,c*; Fun *et al.*, 2010; Jasinski *et al.*, 2010; Baktır *et al.*, 2011).

The crystal structures of some bis-chalcones viz., 2,6-bis(4-methoxybenzylidene)cyclohexanone (Butcher *et al.*, 2006*a*), 1,5-bis(4-chlorophenyl)penta-1,4-dien-3-one (Butcher *et al.*, 2006*b*), 1,5-bis(3,4-dimethoxyphenyl)penta-1,4-dien-3-one (Butcher *et al.*, 2007*a*), 1,5-bis(4-fluorophenyl)penta-1,4-dien-3-one (Butcher *et al.*, 2007*b*), 2,5-bis(3,4-dimethoxybenzylidene)cyclopentanone (Butcher *et al.*, 2007*c*), 1,5-bis(4-methoxyphenyl)penta-1,4-dien-3-one (Harrison *et al.*, 2006), 2,4-dimethyl-1,5-diphenylpenta-1,4-dien-3-one (Hu *et al.*, 2004) have been reported. In continuation of our work on synthesis of chalcone derivatives, the title compound (I) was prepared and its crystal structure is reported.

In the title molecule (I), (Fig. 1), bond lengths and angles are comparable to closely related structures (Patil *et al.*, 2007; Zhao *et al.*, 2007; Fischer *et al.*, 2007; Butcher *et al.* (2006*a,b*, 2007*a,b,c*); Harrison *et al.* (2006); Hu *et al.* (2004). The least-squares plane through the C7–C11/O3 group makes dihedral angles of 8.22 (6) and 32.14 (6)° with the C1–C6 benzene ring and the C12–C17 phenyl ring, respectively. The dihedral angle between these rings is 31.21 (5)° and the nitro group at C3 lies close to the C1–C6 ring plane, with O1–N1–C3–C2 and O2–N1–C3–C4 torsion angles of 6.3 (3) and 7.4 (3)°, respectively.

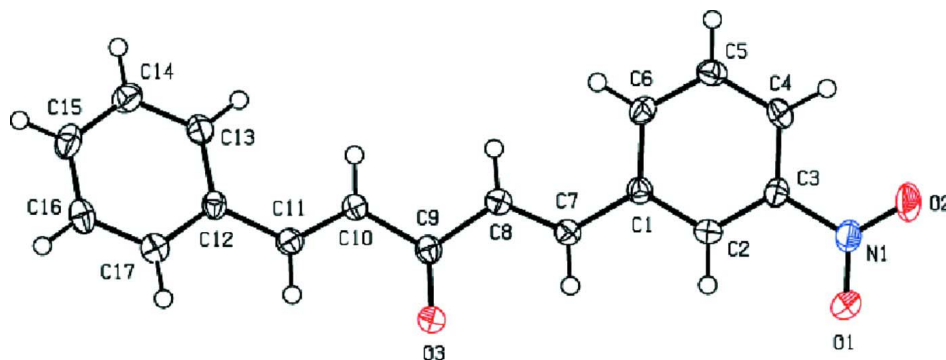
The molecular packing of (I) shown in Fig. 2 is stabilized by C—H \cdots O interactions (Table 1), which lead to the formation of a centrosymmetric dimer, and is further consolidated by C—H \cdots π interactions (Table 1) involving the C12–C17 phenyl ring.

S2. Experimental

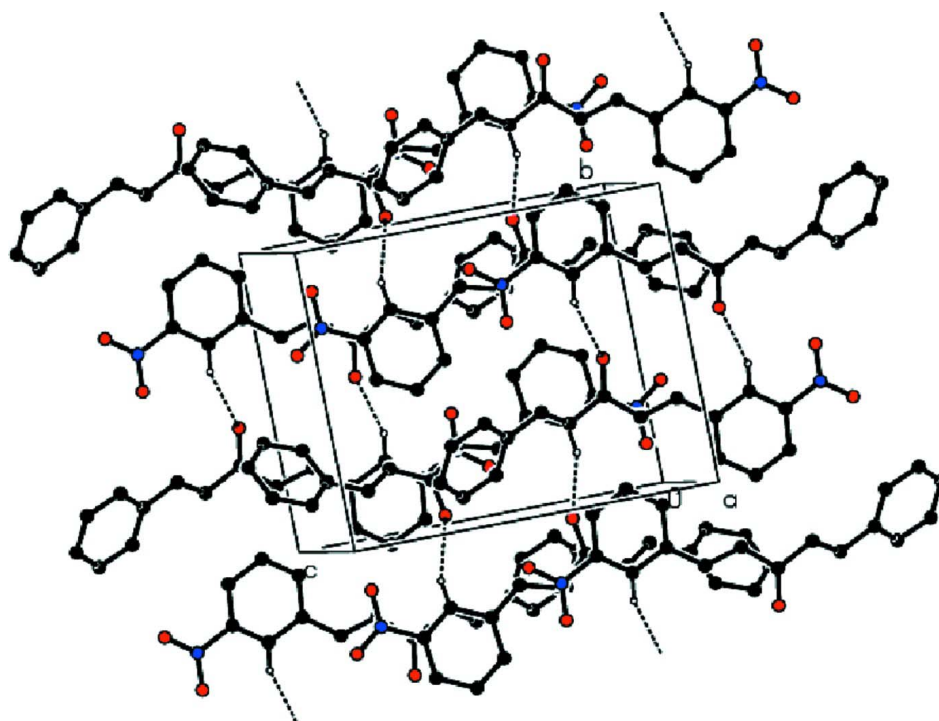
Synthesis of title compound was carried out by stirring a mixture of benzylidene acetone (1.46 g, 0.01 mol) and 3-nitrobenzaldehyde (1.51 g, 0.01 mol) in 40 ml of ethanolic sodium hydroxide at 278–283 K for 3 h. The precipitate was collected by filtration and purified by recrystallization from ethanol. The single crystal was grown from 1,4-dioxane by the slow evaporation method and yield of the compound was 80%. (*M.pt.* 414 K).

S3. Refinement

All H atoms were positioned geometrically (C—H = 0.93 Å) and refined as riding, with $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

**Figure 1**

The molecular structure of the title compound, showing 50% probability displacement ellipsoids and the atom-numbering scheme.

**Figure 2**

The crystal packing of the title compound, viewed approximately along the *a* axis. H atoms not involved in the intermolecular interactions (dashed lines) have been omitted for clarity.

(1*E*,4*E*)-1-(3-Nitrophenyl)-5-phenylpenta-1,4-dien-3-one

Crystal data

$C_{17}H_{13}NO_3$

$M_r = 279.28$

Monoclinic, $P2_1/c$

Hall symbol: $-P\ 2_1/c$

$a = 11.9806\ (6)\ \text{\AA}$

$b = 9.8955\ (4)\ \text{\AA}$

$c = 12.5562\ (7)\ \text{\AA}$

$\beta = 114.992\ (7)^\circ$

$V = 1349.21\ (14)\ \text{\AA}^3$

$Z = 4$

$F(000) = 584$

$D_x = 1.375\ \text{Mg m}^{-3}$

Mo $K\alpha$ radiation, $\lambda = 0.71073\ \text{\AA}$

Cell parameters from 2209 reflections

$\theta = 3.1\text{--}30.9^\circ$
 $\mu = 0.10\text{ mm}^{-1}$
 $T = 293\text{ K}$

Plate, colourless
 $0.44 \times 0.34 \times 0.08\text{ mm}$

Data collection

Oxford Diffraction Xcalibur Ruby Gemini
 diffractometer
 Radiation source: Enhance (Mo) X-ray Source
 Graphite monochromator
 Detector resolution: $10.5081\text{ pixels mm}^{-1}$
 ω scans
 Absorption correction: multi-scan
 (CrysAlis RED; Oxford Diffraction, 2007)
 $T_{\min} = 0.969$, $T_{\max} = 1.000$

7835 measured reflections
 3746 independent reflections
 2453 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.033$
 $\theta_{\max} = 30.9^\circ$, $\theta_{\min} = 3.3^\circ$
 $h = -17 \rightarrow 15$
 $k = -13 \rightarrow 14$
 $l = -11 \rightarrow 18$

Refinement

Refinement on F^2
 Least-squares matrix: full
 $R[F^2 > 2\sigma(F^2)] = 0.055$
 $wR(F^2) = 0.128$
 $S = 1.02$
 3746 reflections
 190 parameters
 0 restraints
 Primary atom site location: structure-invariant
 direct methods

Secondary atom site location: difference Fourier
 map
 Hydrogen site location: inferred from
 neighbouring sites
 H-atom parameters constrained
 $w = 1/[\sigma^2(F_o^2) + (0.047P)^2 + 0.2703P]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\max} < 0.001$
 $\Delta\rho_{\max} = 0.25\text{ e \AA}^{-3}$
 $\Delta\rho_{\min} = -0.26\text{ e \AA}^{-3}$

Special details

Geometry. Bond distances, angles *etc.* have been calculated using the rounded fractional coordinates. All su's are estimated from the variances of the (full) variance-covariance matrix. The cell e.s.d.'s are taken into account in the estimation of distances, angles and torsion angles

Refinement. Refinement on F^2 for ALL reflections except those flagged by the user for potential systematic errors. Weighted R -factors wR and all goodnesses of fit S are based on F^2 , conventional R -factors R are based on F , with F set to zero for negative F^2 . The observed criterion of $F^2 > \sigma(F^2)$ is used only for calculating $-R$ -factor-obs *etc.* and is not relevant to the choice of reflections for refinement. R -factors based on F^2 are statistically about twice as large as those based on F , and R -factors based on ALL data will be even larger.

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters (\AA^2)

	<i>x</i>	<i>y</i>	<i>z</i>	$U_{\text{iso}}^*/U_{\text{eq}}$
O1	0.17385 (11)	0.14120 (12)	0.04941 (10)	0.0346 (4)
O2	0.11627 (15)	0.33639 (14)	−0.03094 (11)	0.0526 (5)
O3	0.54296 (11)	0.04578 (11)	0.67363 (10)	0.0288 (4)
N1	0.16833 (13)	0.26465 (15)	0.05501 (12)	0.0291 (4)
C1	0.34756 (14)	0.30928 (16)	0.37830 (13)	0.0208 (4)
C2	0.29603 (14)	0.25023 (16)	0.26714 (13)	0.0211 (4)
C3	0.22682 (14)	0.32911 (16)	0.17068 (13)	0.0226 (5)
C4	0.20852 (15)	0.46592 (17)	0.17935 (14)	0.0248 (5)
C5	0.26174 (15)	0.52505 (17)	0.28955 (14)	0.0247 (5)
C6	0.32965 (15)	0.44777 (16)	0.38746 (14)	0.0242 (5)
C7	0.41680 (14)	0.22370 (16)	0.48051 (14)	0.0225 (5)
C8	0.46170 (14)	0.26055 (16)	0.59296 (14)	0.0227 (5)
C9	0.53494 (14)	0.16722 (16)	0.68958 (14)	0.0225 (5)

C10	0.60212 (14)	0.23207 (16)	0.80546 (13)	0.0221 (5)
C11	0.70265 (14)	0.17661 (16)	0.88844 (13)	0.0226 (5)
C12	0.78098 (14)	0.23472 (16)	1.00356 (13)	0.0211 (4)
C13	0.74719 (15)	0.34930 (17)	1.04775 (14)	0.0256 (5)
C14	0.82268 (16)	0.39866 (17)	1.15834 (15)	0.0296 (5)
C15	0.93264 (16)	0.33433 (18)	1.22672 (15)	0.0307 (5)
C16	0.96719 (15)	0.22128 (18)	1.18395 (15)	0.0288 (5)
C17	0.89256 (15)	0.17173 (17)	1.07295 (14)	0.0258 (5)
H2A	0.30800	0.15890	0.25790	0.0250*
H4A	0.16180	0.51670	0.11310	0.0300*
H5A	0.25180	0.61710	0.29780	0.0300*
H6A	0.36420	0.48870	0.46100	0.0290*
H7A	0.43070	0.13480	0.46530	0.0270*
H8A	0.44660	0.34770	0.61140	0.0270*
H10A	0.57350	0.31380	0.82090	0.0260*
H11A	0.72550	0.09230	0.87160	0.0270*
H13A	0.67340	0.39290	1.00270	0.0310*
H14A	0.79940	0.47530	1.18680	0.0360*
H15A	0.98280	0.36730	1.30110	0.0370*
H16A	1.04090	0.17790	1.22970	0.0350*
H17A	0.91710	0.09590	1.04460	0.0310*

Atomic displacement parameters (\AA^2)

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
O1	0.0408 (8)	0.0265 (6)	0.0284 (7)	−0.0001 (6)	0.0068 (6)	−0.0062 (6)
O2	0.0800 (11)	0.0377 (8)	0.0184 (7)	−0.0015 (8)	−0.0003 (7)	0.0065 (6)
O3	0.0347 (7)	0.0216 (6)	0.0241 (6)	0.0027 (5)	0.0066 (5)	−0.0001 (5)
N1	0.0322 (8)	0.0298 (8)	0.0206 (7)	−0.0031 (7)	0.0066 (6)	−0.0004 (7)
C1	0.0195 (7)	0.0224 (8)	0.0202 (8)	0.0009 (6)	0.0082 (7)	0.0004 (7)
C2	0.0216 (7)	0.0183 (7)	0.0243 (8)	0.0010 (6)	0.0107 (7)	0.0001 (7)
C3	0.0240 (8)	0.0249 (8)	0.0171 (8)	−0.0039 (7)	0.0070 (7)	−0.0009 (7)
C4	0.0270 (8)	0.0251 (8)	0.0211 (8)	0.0029 (7)	0.0091 (7)	0.0067 (7)
C5	0.0307 (9)	0.0199 (8)	0.0270 (9)	0.0024 (7)	0.0156 (7)	0.0023 (7)
C6	0.0278 (8)	0.0259 (8)	0.0199 (8)	−0.0011 (7)	0.0112 (7)	−0.0032 (7)
C7	0.0216 (8)	0.0207 (8)	0.0226 (8)	0.0013 (7)	0.0068 (7)	−0.0001 (7)
C8	0.0228 (8)	0.0211 (8)	0.0230 (8)	0.0016 (7)	0.0085 (7)	−0.0001 (7)
C9	0.0204 (8)	0.0240 (8)	0.0222 (8)	−0.0003 (7)	0.0081 (7)	0.0012 (7)
C10	0.0253 (8)	0.0207 (8)	0.0192 (8)	0.0012 (7)	0.0084 (7)	0.0014 (7)
C11	0.0268 (8)	0.0206 (8)	0.0215 (8)	−0.0006 (7)	0.0112 (7)	0.0010 (7)
C12	0.0223 (8)	0.0227 (8)	0.0177 (7)	−0.0028 (7)	0.0079 (7)	0.0029 (7)
C13	0.0239 (8)	0.0259 (8)	0.0238 (8)	−0.0001 (7)	0.0071 (7)	0.0027 (7)
C14	0.0334 (9)	0.0262 (9)	0.0274 (9)	−0.0039 (8)	0.0111 (8)	−0.0043 (8)
C15	0.0310 (9)	0.0338 (10)	0.0223 (9)	−0.0092 (8)	0.0064 (8)	−0.0038 (8)
C16	0.0209 (8)	0.0352 (10)	0.0253 (9)	−0.0005 (7)	0.0050 (7)	0.0049 (8)
C17	0.0251 (8)	0.0264 (8)	0.0255 (8)	0.0006 (7)	0.0104 (7)	0.0010 (7)

Geometric parameters (Å, °)

O1—N1	1.2270 (19)	C13—C14	1.387 (2)
O2—N1	1.2202 (19)	C14—C15	1.386 (3)
O3—C9	1.2287 (19)	C15—C16	1.378 (3)
N1—C3	1.465 (2)	C16—C17	1.389 (2)
C1—C2	1.394 (2)	C2—H2A	0.9300
C1—C6	1.399 (2)	C4—H4A	0.9300
C1—C7	1.467 (2)	C5—H5A	0.9300
C2—C3	1.384 (2)	C6—H6A	0.9300
C3—C4	1.383 (2)	C7—H7A	0.9300
C4—C5	1.385 (2)	C8—H8A	0.9300
C5—C6	1.383 (2)	C10—H10A	0.9300
C7—C8	1.332 (2)	C11—H11A	0.9300
C8—C9	1.482 (2)	C13—H13A	0.9300
C9—C10	1.479 (2)	C14—H14A	0.9300
C10—C11	1.333 (2)	C15—H15A	0.9300
C11—C12	1.468 (2)	C16—H16A	0.9300
C12—C13	1.395 (2)	C17—H17A	0.9300
C12—C17	1.397 (2)		
O1—N1—O2	123.25 (14)	C12—C17—C16	120.61 (16)
O1—N1—C3	118.38 (13)	C1—C2—H2A	120.00
O2—N1—C3	118.36 (14)	C3—C2—H2A	120.00
C2—C1—C6	118.32 (14)	C3—C4—H4A	121.00
C2—C1—C7	118.78 (14)	C5—C4—H4A	121.00
C6—C1—C7	122.89 (14)	C4—C5—H5A	120.00
C1—C2—C3	119.22 (15)	C6—C5—H5A	120.00
N1—C3—C2	118.73 (14)	C1—C6—H6A	119.00
N1—C3—C4	118.61 (14)	C5—C6—H6A	119.00
C2—C3—C4	122.65 (14)	C1—C7—H7A	117.00
C3—C4—C5	118.09 (15)	C8—C7—H7A	117.00
C4—C5—C6	120.27 (15)	C7—C8—H8A	119.00
C1—C6—C5	121.43 (15)	C9—C8—H8A	119.00
C1—C7—C8	126.52 (15)	C9—C10—H10A	119.00
C7—C8—C9	122.11 (15)	C11—C10—H10A	119.00
O3—C9—C8	122.43 (14)	C10—C11—H11A	117.00
O3—C9—C10	122.50 (15)	C12—C11—H11A	117.00
C8—C9—C10	115.02 (14)	C12—C13—H13A	120.00
C9—C10—C11	121.79 (15)	C14—C13—H13A	120.00
C10—C11—C12	126.80 (15)	C13—C14—H14A	120.00
C11—C12—C13	122.53 (15)	C15—C14—H14A	120.00
C11—C12—C17	118.99 (15)	C14—C15—H15A	120.00
C13—C12—C17	118.46 (14)	C16—C15—H15A	120.00
C12—C13—C14	120.55 (16)	C15—C16—H16A	120.00
C13—C14—C15	120.36 (16)	C17—C16—H16A	120.00
C14—C15—C16	119.65 (16)	C12—C17—H17A	120.00
C15—C16—C17	120.37 (17)	C16—C17—H17A	120.00

O1—N1—C3—C2	6.3 (3)	C1—C7—C8—C9	−177.47 (17)
O1—N1—C3—C4	−172.40 (17)	C7—C8—C9—O3	−11.3 (3)
O2—N1—C3—C2	−173.88 (18)	C7—C8—C9—C10	166.11 (17)
O2—N1—C3—C4	7.4 (3)	O3—C9—C10—C11	21.9 (3)
C6—C1—C2—C3	−1.6 (3)	C8—C9—C10—C11	−155.48 (17)
C7—C1—C2—C3	177.69 (17)	C9—C10—C11—C12	176.13 (16)
C2—C1—C6—C5	0.6 (3)	C10—C11—C12—C13	10.4 (3)
C7—C1—C6—C5	−178.66 (18)	C10—C11—C12—C17	−170.91 (18)
C2—C1—C7—C8	−173.22 (18)	C11—C12—C13—C14	178.32 (17)
C6—C1—C7—C8	6.1 (3)	C17—C12—C13—C14	−0.3 (3)
C1—C2—C3—N1	−177.19 (16)	C11—C12—C17—C16	−177.90 (16)
C1—C2—C3—C4	1.5 (3)	C13—C12—C17—C16	0.8 (3)
N1—C3—C4—C5	178.43 (17)	C12—C13—C14—C15	−0.3 (3)
C2—C3—C4—C5	−0.2 (3)	C13—C14—C15—C16	0.4 (3)
C3—C4—C5—C6	−0.8 (3)	C14—C15—C16—C17	0.0 (3)
C4—C5—C6—C1	0.6 (3)	C15—C16—C17—C12	−0.7 (3)

Hydrogen-bond geometry (\AA , $^\circ$)

Cg1 is the centroid of the C12–C17 phenyl ring.

$D\cdots H\cdots A$	$D\cdots H$	$H\cdots A$	$D\cdots A$	$D\cdots H\cdots A$
C2—H2A \cdots O3 ⁱ	0.93	2.59	3.412 (2)	147
C7—H7A \cdots O3	0.93	2.54	2.858 (2)	100
C11—H11A \cdots O3	0.93	2.57	2.8724 (19)	100
C5—H5A \cdots Cg1 ⁱⁱ	0.93	2.62	3.3915 (19)	141

Symmetry codes: (i) $-x+1, -y, -z+1$; (ii) $-x+1, y+1/2, -z+3/2$.